



# **CHEM11042 *Fundamentals of Chemistry***

## **Term 1 - 2020**

Profile information current as at 16/05/2024 02:07 am

All details in this unit profile for CHEM11042 have been officially approved by CQUniversity and represent a learning partnership between the University and you (our student). The information will not be changed unless absolutely necessary and any change will be clearly indicated by an approved correction included in the profile.

### Corrections

#### **Unit Profile Correction added on 06-04-20**

The end of term examination has now been changed to an alternate form of assessment. Please see your Moodle site for details of the assessment.

## General Information

### Overview

This unit will provide you with the fundamental principles of chemistry that underpin the medical sciences and provide a strong foundation on which you can develop an understanding of biochemistry and molecular science. You will gain an appreciation of the nature of matter, classic atomic structure and how energy is involved in bond formation. These concepts will be developed to explain the forces between molecules that govern chemical interaction. You will be introduced to the chemistry of electrolytes, acids, bases and buffers. This study will be supported by simple calculations to assist you in relating to the pH scale. The study of organic chemistry and molecules central to the life sciences will enable you to develop an understanding of the biochemistry and molecular biology relevant to your specific discipline. The naming and classifying of chemical compounds will enable you to be conversant with accepted scientific terms. Tutorials and on-line activities will complement the theoretical knowledge gained in lectures and provide you with the basic mathematical and analytical tools required in the application of chemistry to your specific discipline.

### Details

Career Level: *Undergraduate*

Unit Level: *Level 1*

Credit Points: 6

Student Contribution Band: 8

Fraction of Full-Time Student Load: 0.125

### Pre-requisites or Co-requisites

There are no requisites for this unit.

Important note: Students enrolled in a subsequent unit who failed their pre-requisite unit, should drop the subsequent unit before the census date or within 10 working days of Fail grade notification. Students who do not drop the unit in this timeframe cannot later drop the unit without academic and financial liability. See details in the [Assessment Policy and Procedure \(Higher Education Coursework\)](#).

### Offerings For Term 1 - 2020

- Online
- Rockhampton

### Attendance Requirements

All on-campus students are expected to attend scheduled classes – in some units, these classes are identified as a mandatory (pass/fail) component and attendance is compulsory. International students, on a student visa, must maintain a full time study load and meet both attendance and academic progress requirements in each study period (satisfactory attendance for International students is defined as maintaining at least an 80% attendance record).

### Website

[This unit has a website, within the Moodle system, which is available two weeks before the start of term. It is important that you visit your Moodle site throughout the term. Please visit Moodle for more information.](#)

## Class and Assessment Overview

### Recommended Student Time Commitment

Each 6-credit Undergraduate unit at CQUniversity requires an overall time commitment of an average of 12.5 hours of study per week, making a total of 150 hours for the unit.

### Class Timetable

#### [Regional Campuses](#)

Bundaberg, Cairns, Emerald, Gladstone, Mackay, Rockhampton, Townsville

#### [Metropolitan Campuses](#)

Adelaide, Brisbane, Melbourne, Perth, Sydney

### Assessment Overview

#### 1. **Online Quiz(zes)**

Weighting: 30%

#### 2. **Written Assessment**

Weighting: 20%

#### 3. **Examination**

Weighting: 50%

### Assessment Grading

This is a graded unit: your overall grade will be calculated from the marks or grades for each assessment task, based on the relative weightings shown in the table above. You must obtain an overall mark for the unit of at least 50%, or an overall grade of 'pass' in order to pass the unit. If any 'pass/fail' tasks are shown in the table above they must also be completed successfully ('pass' grade). You must also meet any minimum mark requirements specified for a particular assessment task, as detailed in the 'assessment task' section (note that in some instances, the minimum mark for a task may be greater than 50%). Consult the [University's Grades and Results Policy](#) for more details of interim results and final grades.

## CQUniversity Policies

**All University policies are available on the [CQUniversity Policy site](#).**

You may wish to view these policies:

- Grades and Results Policy
- Assessment Policy and Procedure (Higher Education Coursework)
- Review of Grade Procedure
- Student Academic Integrity Policy and Procedure
- Monitoring Academic Progress (MAP) Policy and Procedure – Domestic Students
- Monitoring Academic Progress (MAP) Policy and Procedure – International Students
- Student Refund and Credit Balance Policy and Procedure
- Student Feedback – Compliments and Complaints Policy and Procedure
- Information and Communications Technology Acceptable Use Policy and Procedure

This list is not an exhaustive list of all University policies. The full list of University policies are available on the [CQUniversity Policy site](#).

## Previous Student Feedback

### Feedback, Recommendations and Responses

Every unit is reviewed for enhancement each year. At the most recent review, the following staff and student feedback items were identified and recommendations were made.

#### Feedback from Have your Say

##### **Feedback**

Some students still do not believe that chemistry is relevant to their chosen field, this also impacted negatively on their engagement with the units learning materials.

##### **Recommendation**

Following significant consultation with staff from the appropriate disciplines, the unit was designed to present topics in a relevant, scenario based content.

#### Feedback from Have your Say

##### **Feedback**

A number of students expressed their concern at the amount of content covered in this unit.

##### **Recommendation**

CHEM11042 is a pre-requisite to a number of units and in order to prepare the students for these units it is necessary to cover a lot of chemistry fundamentals. Staff can sympathise that students (particularly those studying chemistry for the first time) may feel overwhelmed and struggle with the unit content. Consequently, a large amount of resources were made available to students and students were made aware of the support facilities available to them such as the ALC and so forth. The unit and its content have seen continued review since its redesign in 2017. Consequently, the unit has been continuously updated, in fact, it should be noted that there was a reduction in the amount of content covered in CHEM11042 compared to previous offerings as certain topics were delivered in other units.

#### Feedback from Have your Say Email Moodle Forums

##### **Feedback**

Students raised some sound issues that occurred during live lectures.

##### **Recommendation**

Once academic staff were alerted to student concerns, academic staff took multiple measures to resolve the sound issues, including seeking out maintenance of an air con unit in the lecture theatre. Academic staff also contacted ITD teaching services to remedy the issue - theatre microphone check requests etc. Ultimately, unfortunately most issues were somewhat beyond the control of academic staff.

#### Feedback from Have your say Email

##### **Feedback**

The majority of students indicated that overall they enjoyed the course, the following aspects were highlighted 1. Delivery of content 2. Approachability of staff and their consistent reliable response to queries 3. Quality of face to face interaction with staff 4. Use of online assessment pieces

##### **Recommendation**

The chemistry team continue our efforts to provide students with the necessary resources to aid their understanding of the unit content and support their study.

## Unit Learning Outcomes

### **On successful completion of this unit, you will be able to:**

1. Apply concepts of atomic structure to explain molecular bonding and nuclear reactivity.
2. Apply chemical concepts to healthcare situations.
3. Identify categories of organic compounds and their potential chemical interactions.
4. Perform basic chemical calculations.

## Alignment of Learning Outcomes, Assessment and Graduate Attributes

 N/A Level	 Introductory Level	 Intermediate Level	 Graduate Level	 Professional Level	 Advanced Level
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## Alignment of Assessment Tasks to Learning Outcomes

Assessment Tasks	Learning Outcomes			
	1	2	3	4
1 - Online Quiz(zes) - 30%	•		•	•
2 - Written Assessment - 20%		•		
3 - Examination - 50%	•		•	•

## Alignment of Graduate Attributes to Learning Outcomes

Graduate Attributes	Learning Outcomes			
	1	2	3	4
1 - Communication	•	•	•	•
2 - Problem Solving	•	•	•	•
3 - Critical Thinking	•	•	•	•
4 - Information Literacy	•	•	•	•
5 - Team Work				
6 - Information Technology Competence				
7 - Cross Cultural Competence				
8 - Ethical practice				
9 - Social Innovation				
10 - Aboriginal and Torres Strait Islander Cultures				

## Alignment of Assessment Tasks to Graduate Attributes

Assessment Tasks	Graduate Attributes									
	1	2	3	4	5	6	7	8	9	10
1 - Online Quiz(zes) - 30%		•	•	•		•				
2 - Written Assessment - 20%	•	•	•	•		•				
3 - Examination - 50%	•	•	•	•						

## Textbooks and Resources

### Textbooks

CHEM11042

#### Supplementary

##### Chemistry ( Paper Text + eBook code )

4th Edition (2018)

Authors: Blackman , Bottle , Schmid , Mocerino & Wille

Wiley

Milton , Queensland , Australia

ISBN: 9780730363286

Binding: Paperback

CHEM11042

#### Supplementary

##### Periodic Table

Edition: 2019 (2019)

Authors: CQUniversity

CQUniversity

Rockhampton , QLD , Australia

Binding: Other

##### Additional Textbook Information

The optional text will be useful in furthering your studies. Copies can be purchased at the CQUni Bookshop here:

<http://bookshop.cqu.edu.au> (search on the Unit code)

[View textbooks at the CQUniversity Bookshop](#)

### IT Resources

**You will need access to the following IT resources:**

- CQUniversity Student Email
- Internet
- Unit Website (Moodle)

## Referencing Style

All submissions for this unit must use the referencing style: [Vancouver](#)

For further information, see the Assessment Tasks.

## Teaching Contacts

**Amie Anastasi** Unit Coordinator

[a.anastasi@cqu.edu.au](mailto:a.anastasi@cqu.edu.au)

## Schedule

### Week 1 - 09 Mar 2020

Module/Topic	Chapter	Events and Submissions/Topic
Introduction to Chemistry		
Matter	Chemistry Foundations Study Guide -	
Atoms	Topics 1, 2 and 3	
Periodic Table		

### Week 2 - 16 Mar 2020

Module/Topic	Chapter	Events and Submissions/Topic
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Ions Formation  
Bonding  
Intermolecular Forces

Chemistry Foundations Study Guide -  
Topics 4, 5, 6 and 7

### Week 3 - 23 Mar 2020

Module/Topic	Chapter	Events and Submissions/Topic
The Mole Mole-Mass Conversions Solutions and Dilutions	Chemistry Foundations Study Guide - Topics 8 and 9	Assessment Item 1 - Online Quiz 1 Quiz closes 11:55 pm (AEST), Sunday March 29, 2020

### Week 4 - 30 Mar 2020

Module/Topic	Chapter	Events and Submissions/Topic
Chemical Equations 1	Chemistry Foundations Study Guide - Topics 10 and 11	

### Week 5 - 06 Apr 2020

Module/Topic	Chapter	Events and Submissions/Topic
Chemical Equations 2	Chemistry Foundations Study Guide - Topics 11 and 12	

### Vacation Week - 13 Apr 2020

Module/Topic	Chapter	Events and Submissions/Topic
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### Week 6 - 20 Apr 2020

Module/Topic	Chapter	Events and Submissions/Topic
Electrolytes Acids and Bases	Chemistry Foundations Study Guide - Topics 13 and 14	

### Week 7 - 27 Apr 2020

Module/Topic	Chapter	Events and Submissions/Topic
Buffers Introduction to Nuclear Chemistry	Chemistry Foundations Study Guide - Topics 15 and 16	Assessment Item 2 - Written Assessment Due: 11:55 pm (AEST), Sunday May 3, 2020

### Week 8 - 04 May 2020

Module/Topic	Chapter	Events and Submissions/Topic
Introduction to Organic Chemistry Saturated Hydrocarbons	Chemistry Foundations Study Guide - Topics 17 and 18	

### Week 9 - 11 May 2020

Module/Topic	Chapter	Events and Submissions/Topic
Unsaturated Hydrocarbons Aromatics	Chemistry Foundations Study Guide - Topics 19 and 20	

### Week 10 - 18 May 2020

Module/Topic	Chapter	Events and Submissions/Topic
Alcohols Thiols Amines	Chemistry Foundations Study Guide - Topics 21, 22 and 23	

### Week 11 - 25 May 2020

Module/Topic	Chapter	Events and Submissions/Topic
Aldehydes and Ketones Carboxylic Acids and their Derivatives	Chemistry Foundations Study Guide - Topics 24 and 25	

### Week 12 - 01 Jun 2020

Module/Topic	Chapter	Events and Submissions/Topic
Review		Assessment Item 1 - Online Quiz 2 Quiz closes 11:55 pm (AEST), Sunday June 7, 2020

**Review/Exam Week - 08 Jun 2020**

Module/Topic	Chapter	Events and Submissions/Topic
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**Exam Week - 15 Jun 2020**

Module/Topic	Chapter	Events and Submissions/Topic
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## Term Specific Information

A Study Guide for this unit will be provided via the Moodle site. This Study Guide will be your reference material for this unit. Please refer to the Textbooks and Resources section of this unit profile for additional textbook requirements.

## Assessment Tasks

### 1 Online Quiz

**Assessment Type**

Online Quiz(zes)

**Task Description**

This assessment is comprised of 2 online quizzes which will assess your understanding of the topics presented in this unit. This assessment requires you to apply the concepts presented in lectures to answer a series of multiple choice questions. All questions in each quiz are of equal value.

Quiz 1 will contribute 10%

Quiz 2 will contribute 20%.

The 2 online quizzes will contribute a total of 30% of the assessment for this unit. The quizzes are not timed and you are allowed two attempts; the highest score of the two attempts will be recorded. Note that quiz questions are generated randomly and you will receive different questions on subsequent attempts.

**Number of Quizzes**

2

**Frequency of Quizzes**

Other

**Assessment Due Date**

Online Quiz 1 be due at 11:55 pm (AEST) on the Sunday of Week 3 (March 29, 2020) and Online Quiz 2 be due at 11:55 pm (AEST) on the Sunday of Week 12 ((June 7, 2020).

**Return Date to Students**

Quizzes will be released after the completion of each attempt. Answers to the quiz questions will be released after each quiz has closed.

**Weighting**

30%

**Assessment Criteria**

All questions are of equal weighting. One mark will be awarded for each correct response. Incorrect responses will not incur a penalty.

**Referencing Style**

- [Vancouver](#)

**Submission**

Online

**Submission Instructions**

Complete each quiz by following the link on the CHEM11042 Moodle site.

**Learning Outcomes Assessed**

- Apply concepts of atomic structure to explain molecular bonding and nuclear reactivity.



- Identify categories of organic compounds and their potential chemical interactions.
- Perform basic chemical calculations.

#### **Graduate Attributes**

- Problem Solving
- Critical Thinking
- Information Literacy
- Information Technology Competence

## **2 Written Assessment**

#### **Assessment Type**

Written Assessment

#### **Task Description**

This assessment is designed to assess your comprehension of the concepts presented in the unit through their application to answer a series of questions.

Marks will be awarded for each question as indicated in the assessment item (please see the Moodle site for further details).

Explanations for each answer must be provided and if calculations are required all workings must be provided.

#### **Assessment Due Date**

The assessment is due at 11:55 pm (AEST) on the Sunday of Week 7 (May 3, 2020)

#### **Return Date to Students**

Week 10 Friday (22 May 2020)

#### **Weighting**

20%

#### **Assessment Criteria**

Marks will be awarded for each question as indicated in the assessment item (please see the Moodle site for further details).

#### **Referencing Style**

- [Vancouver](#)

#### **Submission**

Online

#### **Submission Instructions**

Upload assessment in WORD FORMAT by following instructions on the Moodle site for CHEM11042.

#### **Learning Outcomes Assessed**

- Apply chemical concepts to healthcare situations.

#### **Graduate Attributes**

- Communication
- Problem Solving
- Critical Thinking
- Information Literacy
- Information Technology Competence

## **Examination**

#### **Outline**

Complete an invigilated examination.

#### **Date**

During the examination period at a CQUniversity examination centre.

#### **Weighting**

50%

#### **Length**

120 minutes

#### **Minimum mark or grade**

40%

**Exam Conditions**

Closed Book.

**Materials**

Dictionary - non-electronic, concise, direct translation only (dictionary must not contain any notes or comments).

Calculator - all non-communicable calculators, including scientific, programmable and graphics calculators are authorised

## Academic Integrity Statement

As a CQUniversity student you are expected to act honestly in all aspects of your academic work.

Any assessable work undertaken or submitted for review or assessment must be your own work. Assessable work is any type of work you do to meet the assessment requirements in the unit, including draft work submitted for review and feedback and final work to be assessed.

When you use the ideas, words or data of others in your assessment, you must thoroughly and clearly acknowledge the source of this information by using the correct referencing style for your unit. Using others' work without proper acknowledgement may be considered a form of intellectual dishonesty.

Participating honestly, respectfully, responsibly, and fairly in your university study ensures the CQUniversity qualification you earn will be valued as a true indication of your individual academic achievement and will continue to receive the respect and recognition it deserves.

As a student, you are responsible for reading and following CQUniversity's policies, including the [Student Academic Integrity Policy and Procedure](#). This policy sets out CQUniversity's expectations of you to act with integrity, examples of academic integrity breaches to avoid, the processes used to address alleged breaches of academic integrity, and potential penalties.

**What is a breach of academic integrity?**

A breach of academic integrity includes but is not limited to plagiarism, self-plagiarism, collusion, cheating, contract cheating, and academic misconduct. The Student Academic Integrity Policy and Procedure defines what these terms mean and gives examples.

**Why is academic integrity important?**

A breach of academic integrity may result in one or more penalties, including suspension or even expulsion from the University. It can also have negative implications for student visas and future enrolment at CQUniversity or elsewhere. Students who engage in contract cheating also risk being blackmailed by contract cheating services.

**Where can I get assistance?**

For academic advice and guidance, the [Academic Learning Centre \(ALC\)](#) can support you in becoming confident in completing assessments with integrity and of high standard.

**What can you do to act with integrity?**

**Be Honest**

If your assessment task is done by someone else, it would be dishonest of you to claim it as your own

**Seek Help**

If you are not sure about how to cite or reference in essays, reports etc, then seek help from your lecturer, the library or the Academic Learning Centre (ALC)

**Produce Original Work**

Originality comes from your ability to read widely, think critically, and apply your gained knowledge to address a question or problem