

Profile information current as at 30/04/2024 02:23 am

All details in this unit profile for CHEM11044 have been officially approved by CQUniversity and represent a learning partnership between the University and you (our student). The information will not be changed unless absolutely necessary and any change will be clearly indicated by an approved correction included in the profile.

## **General Information**

## Overview

In this unit, you will learn practical chemistry applications including sample acquisition, chain-of-custody, storage, analysis and reporting. You will become familiar with laboratory compliance procedures, identify risks and appropriate risk-minimisation approaches. The theoretical concepts will include naming of compounds, electrochemistry and its applications, pH, acidity and buffering capacity of solutions, reaction rates and kinetics, organic reaction mechanisms, and basic atomic and molecular spectroscopy. This unit will set a strong foundation for studies in analytical and materials sciences, physical and interface chemistry. You will also be able to examine the energies associated with electrochemical reactions that underpin important processes such as electricity generation, renewable energies, corrosion and electroplating. Accompanying the theory, you will enhance your practical skills by learning the operation and maintenance of common instrumentation in the laboratory, and perform advanced titrations, measure soil acidity, and synthesise, and determine yield and purity of, organic products. On-campus students will attend regular laboratory sessions and there is a compulsory residential school for distance students.

## **Details**

Career Level: Undergraduate

Unit Level: Level 1 Credit Points: 6

Student Contribution Band: 8

Fraction of Full-Time Student Load: 0.125

# Pre-requisites or Co-requisites

Prerequisite: CHEM11043 Atoms, Molecules and Matter

Important note: Students enrolled in a subsequent unit who failed their pre-requisite unit, should drop the subsequent unit before the census date or within 10 working days of Fail grade notification. Students who do not drop the unit in this timeframe cannot later drop the unit without academic and financial liability. See details in the <u>Assessment Policy and Procedure (Higher Education Coursework)</u>.

# Offerings For Term 2 - 2020

Mixed Mode

# Attendance Requirements

All on-campus students are expected to attend scheduled classes – in some units, these classes are identified as a mandatory (pass/fail) component and attendance is compulsory. International students, on a student visa, must maintain a full time study load and meet both attendance and academic progress requirements in each study period (satisfactory attendance for International students is defined as maintaining at least an 80% attendance record).

## Residential Schools

This unit has a Compulsory Residential School for distance mode students and the details are: Click here to see your Residential School Timetable.

## Website

This unit has a website, within the Moodle system, which is available two weeks before the start of term. It is important that you visit your Moodle site throughout the term. Please visit Moodle for more information.

# Class and Assessment Overview

## Recommended Student Time Commitment

Each 6-credit Undergraduate unit at CQUniversity requires an overall time commitment of an average of 12.5 hours of study per week, making a total of 150 hours for the unit.

## Class Timetable

### **Regional Campuses**

Bundaberg, Cairns, Emerald, Gladstone, Mackay, Rockhampton, Townsville

#### **Metropolitan Campuses**

Adelaide, Brisbane, Melbourne, Perth, Sydney

## Assessment Overview

1. Written Assessment

Weighting: 20%

2. Practical Assessment

Weighting: 30% 3. **Take Home Exam** Weighting: 50%

## Assessment Grading

This is a graded unit: your overall grade will be calculated from the marks or grades for each assessment task, based on the relative weightings shown in the table above. You must obtain an overall mark for the unit of at least 50%, or an overall grade of 'pass' in order to pass the unit. If any 'pass/fail' tasks are shown in the table above they must also be completed successfully ('pass' grade). You must also meet any minimum mark requirements specified for a particular assessment task, as detailed in the 'assessment task' section (note that in some instances, the minimum mark for a task may be greater than 50%). Consult the <u>University's Grades and Results Policy</u> for more details of interim results and final grades.

# **CQUniversity Policies**

## All University policies are available on the CQUniversity Policy site.

You may wish to view these policies:

- Grades and Results Policy
- Assessment Policy and Procedure (Higher Education Coursework)
- Review of Grade Procedure
- Student Academic Integrity Policy and Procedure
- Monitoring Academic Progress (MAP) Policy and Procedure Domestic Students
- Monitoring Academic Progress (MAP) Policy and Procedure International Students
- Student Refund and Credit Balance Policy and Procedure
- Student Feedback Compliments and Complaints Policy and Procedure
- Information and Communications Technology Acceptable Use Policy and Procedure

This list is not an exhaustive list of all University policies. The full list of University policies are available on the <u>CQUniversity Policy site</u>.

## Previous Student Feedback

# Feedback, Recommendations and Responses

Every unit is reviewed for enhancement each year. At the most recent review, the following staff and student feedback items were identified and recommendations were made.

## Feedback from Have Your Say

#### **Feedback**

The best aspects of the unit were that lecturers took time to present the more complex aspects of topics in a logical sequence. Tutorial sessions were also well explained which helped students understand the content.

#### Recommendation

We will continue to maintain high standards set going forward.

# Feedback from Have Your Say

#### **Feedback**

There was not enough time in the residential school to complete the experiments and proformas.

#### Recommendation

We have planned to increase the residential school duration from 3 to 4 days in 2020. It is also noted that there wont be any internal laboratory classes scheduled in 2020.

## Feedback from Have Your Say

#### **Feedback**

Learning Resources: Students wanted the most updated version of laboratory manual made available at the beginning of the term.

#### Recommendation

A thorough review of the laboratory manual will be undertaken before 2020 offering. The most updated version will be uploaded on Moodle at the beginning of the term.

# **Unit Learning Outcomes**

## On successful completion of this unit, you will be able to:

- 1. Apply concepts of chemical bonding and reaction energies to chemical synthesis and processes
- 2. Use critical reasoning to apply chemical theories to reactions
- 3. Demonstrate an understanding of buffer chemistry with regard to preparation, buffer capacity and pH
- 4. Synthesise organic compounds and examine these for purity
- 5. Interpret a range of spectra, including infrared, nuclear magnetic resonance and mass spectroscopy, to identify compounds
- 6. Demonstrate competency in laboratory compliance procedures, experimental techniques, data generation, analysis and report writing.

# Alignment of Learning Outcomes, Assessment and Graduate Attributes



# Alignment of Assessment Tasks to Learning Outcomes

Assessment Tasks	Lear	Learning Outcomes				
	1	2	3	4	5	6
1 - Written Assessment - 20%	•	•				

Assessment Tasks		Learning Outcomes								
		1	2		3		4	5		6
2 - Practical Assessment - 30%					•		•	•		•
3 - Take Home Exam - 50%		•	•		•			•		
Alignment of Craduata Attributes to Learni	na Out	- o m								
Alignment of Graduate Attributes to Learning Outcomes  Craduate Attributes  Learning Outcomes										
				1		2	3	4	5	6
1 - Communication				-		-		•		
2 - Problem Solving						•	•	•		
3 - Critical Thinking						•			•	
4 - Information Literacy										
5 - Team Work										
6 - Information Technology Competence										•
7 - Cross Cultural Competence										
8 - Ethical practice										
9 - Social Innovation										
10 - Aboriginal and Torres Strait Islander Cultures										
Alignment of Assessment Tasks to Gradua	te Attrik	oute	es							
Assessment Tasks	Gra	Graduate Attributes								
	1	2	3	4	5	6	7	8	9	10
1 - Written Assessment - 20%	•	•				•				
2 - Practical Assessment - 30%	•	•			•	•				
3 - Take Home Exam - 50%	•	•	•							

# Textbooks and Resources

## **Textbooks**

CHEM11044

#### **Prescribed**

#### Chemistry

Edition: 4th (2019)

Authors: Allan Blackman, Steven Bottle, Siegbert Schmid, Mauro Mocerino, Uta Willie

Wilev

Milton , QLD , Australia ISBN: 0-471-47811-3 Binding: Paperback

## **Additional Textbook Information**

If you prefer to study with a paper copy, they are available at the CQUni Bookshop

here: <a href="http://bookshop.cqu.edu.au">http://bookshop.cqu.edu.au</a> (search on the Unit code). eBooks are available at the publisher's website.

## View textbooks at the CQUniversity Bookshop

## **IT Resources**

## You will need access to the following IT resources:

- CQUniversity Student Email
- Internet
- Unit Website (Moodle)

# Referencing Style

All submissions for this unit must use the referencing style: Vancouver

For further information, see the Assessment Tasks.

# **Teaching Contacts**

Shaneel Chandra Unit Coordinator

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Ty Jones Unit Coordinator

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# Schedule

Week 1 - 13 Jul 2020		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Introduction to Sampling and Spectroscopic Techniques/Chemical Reactions	3, to be advised	
Week 2 - 20 Jul 2020		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Chemical Thermodynamics	8	
Week 3 - 27 Jul 2020		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Chemical Equilibrium	9	

Week 4 - 03 Aug 2020		
Module/Topic	Chapter	Events and Submissions/Topic
Electrochemistry	12	
Week 5 - 10 Aug 2020		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Reaction Kinetics	15	Written Assessment (Short Answer Questions) Due: Week 5 Friday (14 Aug 2020) 11:55 pm AEST
Vacation Week - 17 Aug 2020		
Module/Topic	Chapter	Events and Submissions/Topic
Week 6 - 24 Aug 2020		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Acids and Bases	11	
Week 7 - 31 Aug 2020		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Stereochemistry	17	
Week 8 - 07 Sep 2020		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Reactions of Organic Compounds-Part 1	16, 18	
Week 9 - 14 Sep 2020		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Reactions of Organic Compounds-Part 2	19, 21, 23	
Week 10 - 21 Sep 2020		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Spectroscopy-Part 1	20	
Week 11 - 28 Sep 2020		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Spectroscopy-Part 2	20	
Week 12 - 05 Oct 2020		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Review		
Review/Exam Week - 12 Oct 2020		
Module/Topic	Chapter	Events and Submissions/Topic
Exam Week - 19 Oct 2020		
Module/Topic	Chapter	Events and Submissions/Topic

# **Term Specific Information**

Due to COVID-19 impacts at the time of preparing this unit profile, the residential school attached to this unit for Term 2 2020 has been postponed and will need to be completed at a later date. Further details will be made available on the unit Moodle site in due course.

## **Assessment Tasks**

# 1 Written Assessment (Short Answer Questions)

#### **Assessment Type**

Written Assessment

#### **Task Description**

This assessment will require you to interpret and attempt short answer questions related to content in weeks 1, 2, and 3. You will be required to type your solutions and clearly show any calculations where appropriate. Further detail and instructions will be given on the unit Moodle site.

#### **Assessment Due Date**

Week 5 Friday (14 Aug 2020) 11:55 pm AEST

Submissions via the Moodle site for CHEM11044 and must be in MS Word format

#### **Return Date to Students**

Week 6 Friday (28 Aug 2020)

Two weeks after submission with feedback via Moodle.

#### Weighting

20%

#### **Assessment Criteria**

Marks for each question will be awarded as indicated on the assessment item.

Marks will be awarded for:

- application and explanation of relevant content pertaining to chemical thermodynamics and equilibrium
- relevance and clarity of diagrams where appropriate
- clarity of explanations where appropriate
- correct calculations and use of significant figures and units

## **Referencing Style**

Vancouver

### **Submission**

Online

#### **Submission Instructions**

Submissions must be in Microsoft Word Format on the Moodle Unit site

## **Learning Outcomes Assessed**

- · Apply concepts of chemical bonding and reaction energies to chemical synthesis and processes
- Use critical reasoning to apply chemical theories to reactions

### **Graduate Attributes**

- Communication
- Problem Solving
- Information Technology Competence

## 2 Practical Assessment

## **Assessment Type**

**Practical Assessment** 

#### **Task Description**

During a compulsory residential school, you will complete eight practicals that have been designed to reinforce topics covered in lectures. You will be required to collect results, analyse data, and report your findings in an appropriate manner

Seven of the practicals will be completed using a Profoma that you will fill out. One "Unknown Practical" will be required to be written up as a full scientific laboratory report.

Further detail and instructions will be given on the unit Moodle site.

Due to COVID-19 impact, Residential School details have not yet been finalised. Please see the unit Moodle site for the latest details regarding the Residential School offering.

#### **Assessment Due Date**

Proforma will be due at the end of each practical session. The Unknown Practical report will be due two weeks after a compulsory Residential School. For more details refer to the unit Moodle site CHEM11044

#### **Return Date to Students**

Proforma will be assessed and returned in the following practical session. The Practical Report will be returned two weeks after submission.

#### Weighting

30%

#### **Assessment Criteria**

This task will be assessed in the following ways:

#### **Proformas:**

Clearly outline the results, data interpretation and analysis, including calculations.

Weighting - Each proforma will be worth 3% (21 % in total).

## **Unknown Practical (To be submitted Online):**

Full report clearly describing with logic on how you identified the unknown compounds Weighting – 9%

### **Referencing Style**

Vancouver

#### **Submission**

Online

#### **Submission Instructions**

Submit your work on the unit Moodle - CHEM11044 in MS Word Format

## **Learning Outcomes Assessed**

- · Demonstrate an understanding of buffer chemistry with regard to preparation, buffer capacity and pH
- Synthesise organic compounds and examine these for purity
- Interpret a range of spectra, including infrared, nuclear magnetic resonance and mass spectroscopy, to identify compounds
- Demonstrate competency in laboratory compliance procedures, experimental techniques, data generation, analysis and report writing.

## **Graduate Attributes**

- Communication
- Problem Solving
- Team Work
- Information Technology Competence

## 3 Take Home Assessment

#### **Assessment Type**

Take Home Exam

## **Task Description**

Assessment 3 is a written take-home assessment comprised of a series of questions that will cover the content you have studied during this term. This assessment is designed to assess your comprehension of the concepts and learning outcomes presented in the unit.

#### **Assessment Due Date**

Take home assessment- submission date and time to be advised on Moodle.

## **Return Date to Students**

Three weeks after submissions

### Weighting

50%

## Minimum mark or grade

40%

# Assessment Criteria

Marks will be awarded for each question as indicated in the assessment item. All submissions should be typed and saved

as a word document.

## Weighting

50%

# **Referencing Style**

• <u>Vancouver</u>

## **Submission**

Online

#### **Submission Instructions**

Submissions must be in Microsoft Word Format to the unit Moodle Shell CHEM11044

## **Learning Outcomes Assessed**

- Apply concepts of chemical bonding and reaction energies to chemical synthesis and processes
- Use critical reasoning to apply chemical theories to reactions
- Demonstrate an understanding of buffer chemistry with regard to preparation, buffer capacity and pH
- Interpret a range of spectra, including infrared, nuclear magnetic resonance and mass spectroscopy, to identify compounds

#### **Graduate Attributes**

- Communication
- Problem Solving
- Critical Thinking

# **Academic Integrity Statement**

As a CQUniversity student you are expected to act honestly in all aspects of your academic work.

Any assessable work undertaken or submitted for review or assessment must be your own work. Assessable work is any type of work you do to meet the assessment requirements in the unit, including draft work submitted for review and feedback and final work to be assessed.

When you use the ideas, words or data of others in your assessment, you must thoroughly and clearly acknowledge the source of this information by using the correct referencing style for your unit. Using others' work without proper acknowledgement may be considered a form of intellectual dishonesty.

Participating honestly, respectfully, responsibly, and fairly in your university study ensures the CQUniversity qualification you earn will be valued as a true indication of your individual academic achievement and will continue to receive the respect and recognition it deserves.

As a student, you are responsible for reading and following CQUniversity's policies, including the **Student Academic Integrity Policy and Procedure**. This policy sets out CQUniversity's expectations of you to act with integrity, examples of academic integrity breaches to avoid, the processes used to address alleged breaches of academic integrity, and potential penalties.

## What is a breach of academic integrity?

A breach of academic integrity includes but is not limited to plagiarism, self-plagiarism, collusion, cheating, contract cheating, and academic misconduct. The Student Academic Integrity Policy and Procedure defines what these terms mean and gives examples.

#### Why is academic integrity important?

A breach of academic integrity may result in one or more penalties, including suspension or even expulsion from the University. It can also have negative implications for student visas and future enrolment at CQUniversity or elsewhere. Students who engage in contract cheating also risk being blackmailed by contract cheating services.

### Where can I get assistance?

For academic advice and guidance, the <u>Academic Learning Centre (ALC)</u> can support you in becoming confident in completing assessments with integrity and of high standard.

#### What can you do to act with integrity?



#### **Be Honest**

If your assessment task is done by someone else, it would be dishonest of you to claim it as your own



#### Seek Help

If you are not sure about how to cite or reference in essays, reports etc, then seek help from your lecturer, the library or the Academic Learning Centre (ALC)



### **Produce Original Work**

Originality comes from your ability to read widely, think critically, and apply your gained knowledge to address a question or problem