

Profile information current as at 13/12/2025 03:55 pm

All details in this unit profile for CHEM11044 have been officially approved by CQUniversity and represent a learning partnership between the University and you (our student). The information will not be changed unless absolutely necessary and any change will be clearly indicated by an approved correction included in the profile.

# **General Information**

## Overview

In this unit, you will learn practical chemistry applications including sample acquisition, chain-of-custody, storage, analysis and reporting. You will become familiar with laboratory compliance procedures, identify risks and appropriate risk-minimisation approaches. The theoretical concepts will include naming of compounds, electrochemistry and its applications, pH, acidity and buffering capacity of solutions, reaction rates and kinetics, organic reaction mechanisms, and basic atomic and molecular spectroscopy. This unit will set a strong foundation for studies in analytical and materials sciences, physical and interface chemistry. You will also be able to examine the energies associated with electrochemical reactions that underpin important processes such as electricity generation, renewable energies, corrosion and electroplating. Accompanying the theory, you will enhance your practical skills by learning the operation and maintenance of common instrumentation in the laboratory, and perform advanced titrations, measure soil acidity, and synthesise, and determine yield and purity of, organic products. On-campus students will attend regular laboratory sessions and there is a compulsory residential school for distance students.

# **Details**

Career Level: Undergraduate

Unit Level: Level 1 Credit Points: 6

Student Contribution Band: 8

Fraction of Full-Time Student Load: 0.125

# Pre-requisites or Co-requisites

Prerequisite: CHEM11043 Atoms, Molecules and Matter

Important note: Students enrolled in a subsequent unit who failed their pre-requisite unit, should drop the subsequent unit before the census date or within 10 working days of Fail grade notification. Students who do not drop the unit in this timeframe cannot later drop the unit without academic and financial liability. See details in the <a href="Assessment Policy and Procedure (Higher Education Coursework">Assessment Policy and Procedure (Higher Education Coursework)</a>.

# Offerings For Term 2 - 2022

Mixed Mode

# Attendance Requirements

All on-campus students are expected to attend scheduled classes – in some units, these classes are identified as a mandatory (pass/fail) component and attendance is compulsory. International students, on a student visa, must maintain a full time study load and meet both attendance and academic progress requirements in each study period (satisfactory attendance for International students is defined as maintaining at least an 80% attendance record).

# Website

This unit has a website, within the Moodle system, which is available two weeks before the start of term. It is important that you visit your Moodle site throughout the term. Please visit Moodle for more information.

# Class and Assessment Overview

## Recommended Student Time Commitment

Each 6-credit Undergraduate unit at CQUniversity requires an overall time commitment of an average of 12.5 hours of study per week, making a total of 150 hours for the unit.

# Class Timetable

### **Regional Campuses**

Bundaberg, Cairns, Emerald, Gladstone, Mackay, Rockhampton, Townsville

#### **Metropolitan Campuses**

Adelaide, Brisbane, Melbourne, Perth, Sydney

### **Assessment Overview**

#### 1. Written Assessment

Weighting: 20%

### 2. Practical Assessment

Weighting: 30% 3. **Take Home Exam** Weighting: 50%

# Assessment Grading

This is a graded unit: your overall grade will be calculated from the marks or grades for each assessment task, based on the relative weightings shown in the table above. You must obtain an overall mark for the unit of at least 50%, or an overall grade of 'pass' in order to pass the unit. If any 'pass/fail' tasks are shown in the table above they must also be completed successfully ('pass' grade). You must also meet any minimum mark requirements specified for a particular assessment task, as detailed in the 'assessment task' section (note that in some instances, the minimum mark for a task may be greater than 50%). Consult the <u>University's Grades and Results Policy</u> for more details of interim results and final grades.

# **CQUniversity Policies**

### All University policies are available on the CQUniversity Policy site.

You may wish to view these policies:

- Grades and Results Policy
- Assessment Policy and Procedure (Higher Education Coursework)
- Review of Grade Procedure
- Student Academic Integrity Policy and Procedure
- Monitoring Academic Progress (MAP) Policy and Procedure Domestic Students
- Monitoring Academic Progress (MAP) Policy and Procedure International Students
- Student Refund and Credit Balance Policy and Procedure
- Student Feedback Compliments and Complaints Policy and Procedure
- Information and Communications Technology Acceptable Use Policy and Procedure

This list is not an exhaustive list of all University policies. The full list of University policies are available on the <u>CQUniversity Policy site</u>.

# Previous Student Feedback

# Feedback, Recommendations and Responses

Every unit is reviewed for enhancement each year. At the most recent review, the following staff and student feedback items were identified and recommendations were made.

# Feedback from Have Your Say

#### **Feedback**

Students found the residential school beneficial towards their learning process such as theory and difficult content, topics that were not easy to understand online.

#### Recommendation

The teaching team will continue to maintain the standards and quality of the residential school's activities.

# Feedback from Have Your Say

#### **Feedback**

Whilst the students were very appreciative of the residential school, they also wished to schedule the second group residential school earlier during the term rather than just one week before Take-home exam.

#### Recommendation

Timetabling requests should be considered where possible.

# Feedback from Have Your Say

#### **Feedback**

Students recommended that they would prefer succinct and simple explanations of important concepts of some of the topics in the course.

### Recommendation

The teaching team will work together to develop strategies to provide students with better explanations of important concepts.

# **Unit Learning Outcomes**

## On successful completion of this unit, you will be able to:

- 1. Apply concepts of chemical bonding and reaction energies to chemical synthesis and processes
- 2. Use critical reasoning to apply chemical theories to reactions
- 3. Demonstrate an understanding of buffer chemistry with regard to preparation, buffer capacity and pH
- 4. Synthesise organic compounds and examine these for purity
- 5. Interpret a range of spectra, including infrared, nuclear magnetic resonance and mass spectroscopy, to identify compounds
- 6. Demonstrate competency in laboratory compliance procedures, experimental techniques, data generation, analysis and report writing.

# Alignment of Learning Outcomes, Assessment and Graduate Attributes



# Alignment of Assessment Tasks to Learning Outcomes

Assessment Tasks	Learning Outcomes							
	1	2	3	4	5	6		
1 - Written Assessment - 20%	•	•						

Assessment Tasks		Learning Outcomes								
		1	2		3		4	5		6
2 - Practical Assessment - 30%					•		•	•		•
3 - Take Home Exam - 50%		•	•		•			•		
Alignment of Craduata Attributes to Learni	na Out	- o m								
Alignment of Graduate Attributes to Learning Outcomes  Craduate Attributes  Learning Outcomes										
				1		2	3	4	5	6
1 - Communication				-		-		•		
2 - Problem Solving						•	•	•		
3 - Critical Thinking						•			•	
4 - Information Literacy										
5 - Team Work										
6 - Information Technology Competence										•
7 - Cross Cultural Competence										
8 - Ethical practice										
9 - Social Innovation										
10 - Aboriginal and Torres Strait Islander Cultures										
Alignment of Assessment Tasks to Graduate Attributes										
Assessment Tasks	Gra	Graduate Attributes								
	1	2	3	4	5	6	7	8	9	10
1 - Written Assessment - 20%	•	•				•				
2 - Practical Assessment - 30%	•	•			•	•				
3 - Take Home Exam - 50%	•	•	•							

# Textbooks and Resources

# **Textbooks**

CHEM11044

#### **Prescribed**

#### Chemistry

Edition: 4th (2019)

Authors: Allan Blackman, Steven Bottle, Siegbert Schmid, Mauro Mocerino, Uta Willie

Wilev

Milton , QLD , Australia ISBN: 0-471-47811-3 Binding: Paperback

## **Additional Textbook Information**

This book is available to read online through the Library website. If you would like your own copy, you can purchase either paper or eBook copies at the CQUni Bookshop here: <a href="http://bookshop.cqu.edu.au">http://bookshop.cqu.edu.au</a> (search on the Unit code).

# View textbooks at the CQUniversity Bookshop

# IT Resources

# You will need access to the following IT resources:

- CQUniversity Student Email
- Internet
- Unit Website (Moodle)
- MS Office

# Referencing Style

All submissions for this unit must use the referencing style: <u>Vancouver</u> For further information, see the Assessment Tasks.

# **Teaching Contacts**

Ty Jones Unit Coordinator

t.h.jones@cqu.edu.au

# Schedule

Week 1 - 11 Jul 2022		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Introduction to Sampling and Spectroscopic Techniques/Chemical Reactions	3 and 20	
Week 2 - 18 Jul 2022		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Chemical Thermodynamics	8	
Week 3 - 25 Jul 2022		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Chemical Equilibrium	9	

Week 4 - 01 Aug 2022		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Electrochemistry	12	
Week 5 - 08 Aug 2022		
Module/Topic	Chapter	<b>Events and Submissions/Topic</b>
Reaction Kinetics	15	
Vacation Week - 15 Aug 2022		
Module/Topic	Chapter	Events and Submissions/Topic
Week 6 - 22 Aug 2022		
Module/Topic  Acids and Bases	Chapter 11	Written Assessment (Short Answer Questions)  Due: Week 6, Monday 22 Aug 2022 at 11:55 pm AEST Weighting 20%
Acius anu bases	11	Minimum mark: 50%  Written Assessment (Short Answer Questions) Due: Week 6 Monday (22 Aug 2022) 11:55 pm AEST
Week 7 - 29 Aug 2022		
Module/Topic	Chapter	Events and Submissions/Topic
Stereochemistry	17	
Week 8 - 05 Sep 2022		
Module/Topic	Chapter	Events and Submissions/Topic
Reactions of Organic Compounds-Part 1	16, 18	Residential School (Group A only): From 7 - 10 September 2022. Practical Proformas (7 of them) are due at the beginning of the next day during the Residential School days. Weighting: 15% Minimum mark: 50%
Week 9 - 12 Sep 2022		
Module/Topic	Chapter	1. Residential School (Group B only): From 16 - 19 September 2022.  Practical Proformas (7 of them) are due at the beginning of the next day during the Residential School days.
Reactions of Organic Compounds-Part 2	19, 21, 23	Weighting: 15% Minimum mark: 50%  2. Group A's Unknown Practical Report  Due: Week 9, Sunday 18  September 2022 at 11:55 PM AEST Weighting: 15%  Minimum Mark: 50%
Week 10 - 19 Sep 2022		
Module/Topic	Chapter	Events and Submissions/Topic

**Group B's Unknown Practical** Report Due: Week 11, Monday 18 Spectroscopy-Part 1 20 September 2022 at 11:55 PM AEST Weighting: 15% Minimum Mark: 50% Week 11 - 26 Sep 2022 Module/Topic Chapter **Events and Submissions/Topic** Spectroscopy-Part 2 20 Week 12 - 03 Oct 2022 Module/Topic Chapter **Events and Submissions/Topic** Review Review/Exam Week - 10 Oct 2022 Module/Topic Chapter **Events and Submissions/Topic** Exam Week - 13 Oct 2022 Module/Topic Chapter **Events and Submissions/Topic** Exam Week - 13 Oct 2022 Module/Topic Chapter **Events and Submissions/Topic Assessment 3:** Take-Home Exam Due: Saturday, 15 October 2022 at 11:55 PM AEST Weighting: 50% Minimum mark: 50%

# Term Specific Information

- Four days of Residential Schools at North Rockhampton campus from **7th to 10th September 2022 for Group**A and **16th to 19th September 2022 for Group B** are mandatory for all students.
- Students are required to do self-enrollment for **ONE** of these groups on Moodle.
- Live (Zoom) lecture is available every Monday from 13:00 PM to 14:00 PM during teaching weeks.
- Live tutorial is available every *Tuesday from 14:00 PM to 15:00 PM* during teaching weeks.
- One-to-one consulting time with the Unit Coordinator is available every Wednesday from 10:00 AM to 11:00 AM during teaching weeks.

# **Assessment Tasks**

# 1 Written Assessment (Short Answer Questions)

## **Assessment Type**

Written Assessment

#### **Task Description**

This assessment will require you to interpret and attempt short answer questions related to content in weeks 1, 2, 3, 4, and 5. All submissions should be typed or handwritten and scanned and saved as a word document or PDF file. You are required to show all calculations where appropriate. You will lose your marks if you only provide final answers to questions. Further detail and instructions will be given on the unit Moodle site.

# **Assessment Due Date**

Week 6 Monday (22 Aug 2022) 11:55 pm AEST

Submit your work by the due date on the unit Moodle - CHEM11044 in MS Word Format or PDF File.

#### **Return Date to Students**

Week 8 Friday (9 Sept 2022)

Two weeks after submission with feedback file via Moodle.

### Weighting

20%

#### Minimum mark or grade

50%

#### **Assessment Criteria**

Marks for each question will be awarded as indicated on the assessment item.

Marks will be awarded for:

- application and explanation of relevant content pertaining to chemical thermodynamics and equilibrium
- relevance and clarity of diagrams where appropriate
- clarity of explanations where appropriate
- correct calculations and use of significant figures and units

### **Referencing Style**

• Vancouver

#### **Submission**

Online

#### **Submission Instructions**

Submit your work by the due date on the unit Moodle - CHEM11044 in MS Word Format or PDF file. A late penalty will apply for late submissions and without approval for extension.

#### **Learning Outcomes Assessed**

- Apply concepts of chemical bonding and reaction energies to chemical synthesis and processes
- Use critical reasoning to apply chemical theories to reactions

#### **Graduate Attributes**

- Communication
- Problem Solving
- Information Technology Competence

# 2 Practical Assessment - Unknowns Report

#### **Assessment Type**

**Practical Assessment** 

## **Task Description**

During a compulsory residential school, you will complete eight practicals that have been designed to reinforce topics covered in lectures. You will be required to collect results, analyse data, and report your findings in an appropriate manner.

Seven of the practicals will be completed using a Proforma that you will fill out. One "Unknowns Practical" will be required to be written up as a full scientific laboratory report.

Further detail and instructions will be given on the unit Moodle site.

Please see the unit Moodle site for the latest details regarding the Residential Schol offering.

### Assessment Due Date

Proforma will be due at the beginning of the next day of the practical session. The Unknown Practical report will be due one weeks after a compulsory Residential School. For more details refer to the unit Moodle site CHEM11044

### **Return Date to Students**

Week 12 Friday (7 Oct 2022)

Proforma will be assessed and returned with feedback in the following practical session. The Practical Report will be returned two weeks after submission with feedback file via Moodle..

#### Weighting

30%

## Minimum mark or grade

50%

#### **Assessment Criteria**

This task will be assessed in the following ways:

#### **Proformas:**

Clearly outline the results, data interpretation and analysis, including calculations.

Weighting - 15%

#### **Unknown Practical (To be submitted Online):**

A complete report clearly describing with logic on how you identified the unknown compounds and following the marking quidelines.

Weighting - 15%

## **Referencing Style**

• Vancouver

#### **Submission**

Online

#### **Submission Instructions**

Submit your work by the due date on the unit Moodle - CHEM11044 in MS Word Format or a PDF File. A late penalty will apply for late submissions and without approval for extension.

## **Learning Outcomes Assessed**

- Demonstrate an understanding of buffer chemistry with regard to preparation, buffer capacity and pH
- Synthesise organic compounds and examine these for purity
- Interpret a range of spectra, including infrared, nuclear magnetic resonance and mass spectroscopy, to identify compounds
- Demonstrate competency in laboratory compliance procedures, experimental techniques, data generation, analysis and report writing.

#### **Graduate Attributes**

- Communication
- Problem Solving
- Team Work
- Information Technology Competence

### 3 Take Home Assessment

### **Assessment Type**

Take Home Exam

# **Task Description**

Assessment 3 is a written take-home assessment comprised of a series of questions that will cover the content you have studied during this term. This assessment is designed to assess your comprehension of the concepts and learning outcomes presented in the unit.

#### **Assessment Due Date**

The Take-Home Assessment Questions will be available on Moodle for students during the first week of exam weeks. This assessment will be available for 48 hours ONLY and should be submitted on Moodle - CHEM11044 in MS Word Format or a PDF File. More details about the assessment date and due date will be provided later on Moodle.

#### **Return Date to Students**

Two weeks after submissions with feedback file via Moodle.

# Weighting

50%

#### Minimum mark or grade

50%

#### **Assessment Criteria**

Marks will be awarded for each question as indicated in the assessment item. All submissions should be typed and saved as a word document or a PDF file.

#### Weighting

50%

## **Referencing Style**

• Vancouver

#### **Submission**

Online

#### **Submission Instructions**

Submit your work by the due date on the unit Moodle - CHEM11044 in MS Word Format or PDF File. A late penalty will apply for late submissions and without approval for extension.

# **Learning Outcomes Assessed**

- · Apply concepts of chemical bonding and reaction energies to chemical synthesis and processes
- Use critical reasoning to apply chemical theories to reactions
- Demonstrate an understanding of buffer chemistry with regard to preparation, buffer capacity and pH
- Interpret a range of spectra, including infrared, nuclear magnetic resonance and mass spectroscopy, to identify compounds

### **Graduate Attributes**

- Communication
- Problem Solving
- Critical Thinking

# **Academic Integrity Statement**

As a CQUniversity student you are expected to act honestly in all aspects of your academic work.

Any assessable work undertaken or submitted for review or assessment must be your own work. Assessable work is any type of work you do to meet the assessment requirements in the unit, including draft work submitted for review and feedback and final work to be assessed.

When you use the ideas, words or data of others in your assessment, you must thoroughly and clearly acknowledge the source of this information by using the correct referencing style for your unit. Using others' work without proper acknowledgement may be considered a form of intellectual dishonesty.

Participating honestly, respectfully, responsibly, and fairly in your university study ensures the CQUniversity qualification you earn will be valued as a true indication of your individual academic achievement and will continue to receive the respect and recognition it deserves.

As a student, you are responsible for reading and following CQUniversity's policies, including the **Student Academic Integrity Policy and Procedure**. This policy sets out CQUniversity's expectations of you to act with integrity, examples of academic integrity breaches to avoid, the processes used to address alleged breaches of academic integrity, and potential penalties.

### What is a breach of academic integrity?

A breach of academic integrity includes but is not limited to plagiarism, self-plagiarism, collusion, cheating, contract cheating, and academic misconduct. The Student Academic Integrity Policy and Procedure defines what these terms mean and gives examples.

#### Why is academic integrity important?

A breach of academic integrity may result in one or more penalties, including suspension or even expulsion from the University. It can also have negative implications for student visas and future enrolment at CQUniversity or elsewhere. Students who engage in contract cheating also risk being blackmailed by contract cheating services.

### Where can I get assistance?

For academic advice and guidance, the <u>Academic Learning Centre (ALC)</u> can support you in becoming confident in completing assessments with integrity and of high standard.

#### What can you do to act with integrity?



#### **Be Honest**

If your assessment task is done by someone else, it would be dishonest of you to claim it as your own



#### Seek Help

If you are not sure about how to cite or reference in essays, reports etc, then seek help from your lecturer, the library or the Academic Learning Centre (ALC)



### **Produce Original Work**

Originality comes from your ability to read widely, think critically, and apply your gained knowledge to address a question or problem