



# CHEM12079 *Non-carbon Chemistry*

## Term 1 - 2020

Profile information current as at 18/05/2024 02:35 pm

All details in this unit profile for CHEM12079 have been officially approved by CQUniversity and represent a learning partnership between the University and you (our student). The information will not be changed unless absolutely necessary and any change will be clearly indicated by an approved correction included in the profile.

### Corrections

#### Unit Profile Correction added on 19-04-20

The end of term examination has now been changed to an alternate form of assessment. Please see your Moodle site for details of the assessment.

### General Information

#### Overview

In this unit, you will advance your theoretical knowledge of chemical synthesis with a focus on the structural aspects and energy requirements of bond formation. You will study lattice structures and layers and predict the geometry and bonding properties of molecules using Valence Bond and Molecular Orbital Theories. You will examine the unique properties of the D-block elements and the formation of coordination compounds. You will study interfacial chemistry, for example at solid-liquid boundaries, which lay the foundation for solute transport and is key element of chemical kinetics. Completing this unit will significantly advance your standing as an inorganic chemist and prepare you for the study of advanced topics such as nanotechnology, analytical spectroscopy and materials development.

#### Details

Career Level: *Undergraduate*

Unit Level: *Level 2*

Credit Points: 6

Student Contribution Band: 8

Fraction of Full-Time Student Load: 0.125

#### Pre-requisites or Co-requisites

Pre-requisites: CHEM11044 Chemical Reactions OR CHEM11045 Chemical Investigation and Theory or permission of Head of Course

Important note: Students enrolled in a subsequent unit who failed their pre-requisite unit, should drop the subsequent unit before the census date or within 10 working days of Fail grade notification. Students who do not drop the unit in this timeframe cannot later drop the unit without academic and financial liability. See details in the [Assessment Policy and Procedure \(Higher Education Coursework\)](#).

#### Offerings For Term 1 - 2020

- Mixed Mode

#### Attendance Requirements

All on-campus students are expected to attend scheduled classes – in some units, these classes are identified as a mandatory (pass/fail) component and attendance is compulsory. International students, on a student visa, must maintain a full time study load and meet both attendance and academic progress requirements in each study period (satisfactory attendance for International students is defined as maintaining at least an 80% attendance record).

#### Website

[This unit has a website, within the Moodle system, which is available two weeks before the start of term. It is important that you visit your Moodle site throughout the term. Please visit Moodle for more information.](#)

## Class and Assessment Overview

### Recommended Student Time Commitment

Each 6-credit Undergraduate unit at CQUniversity requires an overall time commitment of an average of 12.5 hours of study per week, making a total of 150 hours for the unit.

### Class Timetable

#### [Regional Campuses](#)

Bundaberg, Cairns, Emerald, Gladstone, Mackay, Rockhampton, Townsville

#### [Metropolitan Campuses](#)

Adelaide, Brisbane, Melbourne, Perth, Sydney

### Assessment Overview

#### 1. **Written Assessment**

Weighting: 25%

#### 2. **Written Assessment**

Weighting: 25%

#### 3. **Examination**

Weighting: 50%

### Assessment Grading

This is a graded unit: your overall grade will be calculated from the marks or grades for each assessment task, based on the relative weightings shown in the table above. You must obtain an overall mark for the unit of at least 50%, or an overall grade of 'pass' in order to pass the unit. If any 'pass/fail' tasks are shown in the table above they must also be completed successfully ('pass' grade). You must also meet any minimum mark requirements specified for a particular assessment task, as detailed in the 'assessment task' section (note that in some instances, the minimum mark for a task may be greater than 50%). Consult the [University's Grades and Results Policy](#) for more details of interim results and final grades.

## CQUniversity Policies

**All University policies are available on the [CQUniversity Policy site](#).**

You may wish to view these policies:

- Grades and Results Policy
- Assessment Policy and Procedure (Higher Education Coursework)
- Review of Grade Procedure
- Student Academic Integrity Policy and Procedure
- Monitoring Academic Progress (MAP) Policy and Procedure – Domestic Students
- Monitoring Academic Progress (MAP) Policy and Procedure – International Students
- Student Refund and Credit Balance Policy and Procedure
- Student Feedback – Compliments and Complaints Policy and Procedure
- Information and Communications Technology Acceptable Use Policy and Procedure

This list is not an exhaustive list of all University policies. The full list of University policies are available on the [CQUniversity Policy site](#).

## Unit Learning Outcomes

**On successful completion of this unit, you will be able to:**

1. Predict the chemistry of coordination compounds as a result of the electronic structure
2. Analyse and compare electron transitions in molecules and compounds using existing theories
3. Relate the unique properties of the D-Block elements and characteristics of coordination compounds to their atomic structure
4. Compare solute transport processes at the liquid interface, and their influences on chemical reactions.










## Alignment of Learning Outcomes, Assessment and Graduate Attributes

 N/A Level	 Introductory Level	 Intermediate Level	 Graduate Level	 Professional Level	 Advanced Level
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### Alignment of Assessment Tasks to Learning Outcomes

Assessment Tasks	Learning Outcomes			
	1	2	3	4
1 - Written Assessment - 25%	•		•	
2 - Written Assessment - 25%		•		•
3 - Examination - 50%	•	•	•	•

### Alignment of Graduate Attributes to Learning Outcomes

Graduate Attributes	Learning Outcomes			
	1	2	3	4
1 - Communication				
2 - Problem Solving				
3 - Critical Thinking				
4 - Information Literacy				
5 - Team Work				
6 - Information Technology Competence				
7 - Cross Cultural Competence				
8 - Ethical practice				
9 - Social Innovation				
10 - Aboriginal and Torres Strait Islander Cultures				

### Alignment of Assessment Tasks to Graduate Attributes

Assessment Tasks	Graduate Attributes									
	1	2	3	4	5	6	7	8	9	10
1 - Written Assessment - 25%										
2 - Written Assessment - 25%										

Assessment Tasks	Graduate Attributes									
	1	2	3	4	5	6	7	8	9	10
<b>3 - Examination - 50%</b>										

## Textbooks and Resources

### Textbooks

CHEM12079

#### Prescribed

#### Inorganic Chemistry

Edition: 5 (2018)

Authors: Catherine E. Housecroft, Alan G. Sharpe

Pearson Education Limited

Binding: Paperback

#### Additional Textbook Information

Copies can be purchased at the CQUni Bookshop here: <http://bookshop.cqu.edu.au> (search on the Unit code)

[View textbooks at the CQUniversity Bookshop](#)

### IT Resources

**You will need access to the following IT resources:**

- CQUniversity Student Email
- Internet
- Unit Website (Moodle)

## Referencing Style

All submissions for this unit must use the referencing style: [Vancouver](#)

For further information, see the Assessment Tasks.

## Teaching Contacts

**Catherine Jones** Unit Coordinator

[c.e.jones@cqu.edu.au](mailto:c.e.jones@cqu.edu.au)

## Schedule

### Week 1 - 09 Mar 2020

Module/Topic	Chapter	Events and Submissions/Topic
Introduction and revision of inorganic chemistry concepts	1, 2, 3.1-3.6	

### Week 2 - 16 Mar 2020

Module/Topic	Chapter	Events and Submissions/Topic
Bonding in polyatomic molecules	5	

### Week 3 - 23 Mar 2020

Module/Topic	Chapter	Events and Submissions/Topic
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Spectroscopic properties of compounds and the theory behind spectroscopic techniques	3.7-3.8; 4.3, 4.5 and 4.7
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#### Week 4 - 30 Mar 2020

Module/Topic	Chapter	Events and Submissions/Topic
Structures and energetics of metallic and ionic solids	6	

#### Week 5 - 06 Apr 2020

Module/Topic	Chapter	Events and Submissions/Topic
Redox chemistry ( $E^\circ$ values; thermodynamics of redox reactions)	8	<b>Written Assessment 1 - Short Answer Questions and Research Question</b> Due: Week 5 Friday (10 Apr 2020) 12:00 pm AEST

#### Vacation Week - 13 Apr 2020

Module/Topic	Chapter	Events and Submissions/Topic
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#### Week 6 - 20 Apr 2020

Module/Topic	Chapter	Events and Submissions/Topic
Descriptive chemistry of hydrogen, s-block and p-block elements - a general overview	10-18 (sections to be advised)	

#### Week 7 - 27 Apr 2020

Module/Topic	Chapter	Events and Submissions/Topic
d-block metal chemistry and the principles of coordination chemistry	19.1-20.4	

#### Week 8 - 04 May 2020

Module/Topic	Chapter	Events and Submissions/Topic
Advanced coordination chemistry	20.6-20.13	

#### Week 9 - 11 May 2020

Module/Topic	Chapter	Events and Submissions/Topic
d-Block metal chemistry: the first row metals	21	

#### Week 10 - 18 May 2020

Module/Topic	Chapter	Events and Submissions/Topic
d-Block metal chemistry: the heavier metals	22	

#### Week 11 - 25 May 2020

Module/Topic	Chapter	Events and Submissions/Topic
Organometallic compounds of d-block elements	24	<b>Written Assessment 2 - Problem Solving and Interpretation</b> Due: Week 11 Monday (25 May 2020) 12:00 pm AEST

#### Week 12 - 01 Jun 2020

Module/Topic	Chapter	Events and Submissions/Topic
Organometallic compounds of s and p-block elements/ review of all content	23	

#### Review/Exam Week - 08 Jun 2020

Module/Topic	Chapter	Events and Submissions/Topic
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## Assessment Tasks

### 1 Written Assessment 1 - Short Answer Questions and Research Question

**Assessment Type**

Written Assessment

**Task Description**

In Written Assessment 1 you will be required to:

1. Answer a series of short answer questions to demonstrate your knowledge of electron transitions and bonding in polyatomic molecules
2. Research the terms: interfacial chemistry, chemical reactions at interfaces, solute transport
3. Locate two recent examples from the literature that have addressed the concepts listed at step 2
4. Summarise the major findings of the journal articles you located at step 3

More information on this assessment task will be made available on Moodle

**Assessment Due Date**

Week 5 Friday (10 Apr 2020) 12:00 pm AEST

**Return Date to Students**

Week 7 Monday (27 Apr 2020)

Assessment marks and feedback will be returned via Moodle on the specified return date

**Weighting**

25%

**Minimum mark or grade**

40

**Assessment Criteria****Short answer questions**

40% of total marks for this assessment

Marks will be awarded for:

- Correctly addressing the questions (20 marks)
- Demonstrating understanding of the key concepts (20 marks)

**Research question**

60% of total marks for this assessment

Marks will be awarded for:

- Quality, depth and accuracy of research (20 marks)
- Demonstrated understanding of the concepts (30 marks)
- Correct referencing, including in-text (10 marks)

More information on this assessment task will be made available on Moodle

**Referencing Style**

- [Vancouver](#)

**Submission**

Online

**Learning Outcomes Assessed**

- Analyse and compare electron transitions in molecules and compounds using existing theories
- Compare solute transport processes at the liquid interface, and their influences on chemical reactions.

**Graduate Attributes**

- Problem Solving

- Critical Thinking

## 2 Written Assessment 2 - Problem Solving and Interpretation

### Assessment Type

Written Assessment

### Task Description

In Written Assessment 2 you will be required to:

1. Complete a series of problem-solving questions in order to:
    - Evaluate and interpret information related to D-block elements and coordination compounds
    - Accurately predict the chemistry of coordination compounds
    - Examine the electron distribution within D-block elements
    - Predict stable oxidation states and associate these to physical properties
  2. Conduct research to locate credible information to support the predictions and interpretations made in step 1.
- More information on this assessment task will be made available on Moodle

### Assessment Due Date

Week 11 Monday (25 May 2020) 12:00 pm AEST

### Return Date to Students

Week 12 Friday (5 June 2020)

Assessments will be returned via Moodle on the specified return date

### Weighting

25%

### Minimum mark or grade

40%

### Assessment Criteria

#### Problem solving and interpretation

Marks will be awarded for:

- Correct evaluation and interpretation of data (30 Marks)
- Demonstrated understanding of the key concepts (40 Marks)
- Quality research to support predictions (20 Marks)
- Correct referencing, including in-text (10 Marks)

More information on this assessment task will be made available on Moodle

### Referencing Style

- [Vancouver](#)

### Submission

Online

### Learning Outcomes Assessed

- Predict the chemistry of coordination compounds as a result of the electronic structure
- Relate the unique properties of the D-Block elements and characteristics of coordination compounds to their atomic structure

### Graduate Attributes

- Problem Solving

## Examination

### Outline

Complete an invigilated examination.

### Date

During the examination period at a CQUniversity examination centre.

### Weighting

50%

### Length

120 minutes

**Minimum mark or grade**

40

**Exam Conditions**

Closed Book.

**Materials**

Dictionary - non-electronic, concise, direct translation only (dictionary must not contain any notes or comments).  
Calculator - all non-communicable calculators, including scientific, programmable and graphics calculators are authorised

## Academic Integrity Statement

As a CQUniversity student you are expected to act honestly in all aspects of your academic work.

Any assessable work undertaken or submitted for review or assessment must be your own work. Assessable work is any type of work you do to meet the assessment requirements in the unit, including draft work submitted for review and feedback and final work to be assessed.

When you use the ideas, words or data of others in your assessment, you must thoroughly and clearly acknowledge the source of this information by using the correct referencing style for your unit. Using others' work without proper acknowledgement may be considered a form of intellectual dishonesty.

Participating honestly, respectfully, responsibly, and fairly in your university study ensures the CQUniversity qualification you earn will be valued as a true indication of your individual academic achievement and will continue to receive the respect and recognition it deserves.

As a student, you are responsible for reading and following CQUniversity's policies, including the [Student Academic Integrity Policy and Procedure](#). This policy sets out CQUniversity's expectations of you to act with integrity, examples of academic integrity breaches to avoid, the processes used to address alleged breaches of academic integrity, and potential penalties.

**What is a breach of academic integrity?**

A breach of academic integrity includes but is not limited to plagiarism, self-plagiarism, collusion, cheating, contract cheating, and academic misconduct. The Student Academic Integrity Policy and Procedure defines what these terms mean and gives examples.

**Why is academic integrity important?**

A breach of academic integrity may result in one or more penalties, including suspension or even expulsion from the University. It can also have negative implications for student visas and future enrolment at CQUniversity or elsewhere. Students who engage in contract cheating also risk being blackmailed by contract cheating services.

**Where can I get assistance?**

For academic advice and guidance, the [Academic Learning Centre \(ALC\)](#) can support you in becoming confident in completing assessments with integrity and of high standard.

**What can you do to act with integrity?**



**Be Honest**

If your assessment task is done by someone else, it would be dishonest of you to claim it as your own

**Seek Help**

If you are not sure about how to cite or reference in essays, reports etc, then seek help from your lecturer, the library or the Academic Learning Centre (ALC)

**Produce Original Work**

Originality comes from your ability to read widely, think critically, and apply your gained knowledge to address a question or problem