



# CHEM40079 *Introductory Chemistry*

## Term 2 - 2022

Profile information current as at 02/10/2022 12:37 pm

All details in this unit profile for CHEM40079 have been officially approved by CQUniversity and represent a learning partnership between the University and you (our student). The information will not be changed unless absolutely necessary and any change will be clearly indicated by an approved correction included in the profile.

## General Information

### Overview

Introductory Chemistry will provide you with a foundation for university study in the chemical sciences. This unit will provide fundamental principles of chemistry by introducing topics such as matter, molecules, bonding, chemical reactions, stoichiometry, acids and bases, and organic chemistry. In addition to gaining an introductory understanding of chemical concepts, you will learn to perform chemical calculations.

### Details

Career Level: *Non-award*

Credit Points: 6

Student Contribution Band: 8

Fraction of Full-Time Student Load: 0.125

### Pre-requisites or Co-requisites

There are no requisites for this unit.

Important note: Students enrolled in a subsequent unit who failed their pre-requisite unit, should drop the subsequent unit before the census date or within 10 working days of Fail grade notification. Students who do not drop the unit in this timeframe cannot later drop the unit without academic and financial liability. See details in the [Assessment Policy and Procedure \(Higher Education Coursework\)](#).

### Offerings For Term 2 - 2022

- Online

### Attendance Requirements

All on-campus students are expected to attend scheduled classes - in some units, these classes are identified as a mandatory (pass/fail) component and attendance is compulsory. International students, on a student visa, must maintain a full time study load and meet both attendance and academic progress requirements in each study period (satisfactory attendance for International students is defined as maintaining at least an 80% attendance record).

### Website

[This unit has a website, within the Moodle system, which is available two weeks before the start of term. It is important that you visit your Moodle site throughout the term. Please visit Moodle for more information.](#)

## Class and Assessment Overview

### Recommended Student Time Commitment

Each 6-credit Non-award unit at CQUniversity requires an overall time commitment of an average of 12.5 hours of study per week, making a total of 150 hours for the unit.

### Class Timetable

#### [Regional Campuses](#)

Bundaberg, Cairns, Emerald, Gladstone, Mackay, Rockhampton, Townsville

#### [Metropolitan Campuses](#)

Adelaide, Brisbane, Melbourne, Perth, Sydney

### Assessment Overview

#### 1. **Online Quiz(zes)**

Weighting: 20%

#### 2. **Written Assessment**

Weighting: 30%

#### 3. **Written Assessment**

Weighting: 50%

### Assessment Grading

This is a graded unit: your overall grade will be calculated from the marks or grades for each assessment task, based on the relative weightings shown in the table above. You must obtain an overall mark for the unit of at least 50%, or an overall grade of 'pass' in order to pass the unit. If any 'pass/fail' tasks are shown in the table above they must also be completed successfully ('pass' grade). You must also meet any minimum mark requirements specified for a particular assessment task, as detailed in the 'assessment task' section (note that in some instances, the minimum mark for a task may be greater than 50%). Consult the [University's Grades and Results Policy](#) for more details of interim results and final grades.

## CQUniversity Policies

**All University policies are available on the [CQUniversity Policy site](#).**

You may wish to view these policies:

- Grades and Results Policy
- Assessment Policy and Procedure (Higher Education Coursework)
- Review of Grade Procedure
- Student Academic Integrity Policy and Procedure
- Monitoring Academic Progress (MAP) Policy and Procedure – Domestic Students
- Monitoring Academic Progress (MAP) Policy and Procedure – International Students
- Student Refund and Credit Balance Policy and Procedure
- Student Feedback – Compliments and Complaints Policy and Procedure
- Information and Communications Technology Acceptable Use Policy and Procedure

This list is not an exhaustive list of all University policies. The full list of University policies are available on the [CQUniversity Policy site](#).

## Previous Student Feedback

### Feedback, Recommendations and Responses

Every unit is reviewed for enhancement each year. At the most recent review, the following staff and student feedback items were identified and recommendations were made.

#### Feedback from Students

##### Feedback

The live lecture/tutorial was well received by students and they liked the opportunity to ask questions.

##### Recommendation

Continue to offer online lectures/tutorials.

#### Feedback from Students

##### Feedback

Students appreciated the personalised Zoom session which provided them with additional support as needed.

##### Recommendation

Continue to offer one on one Zoom sessions.

## Unit Learning Outcomes

### On successful completion of this unit, you will be able to:

1. Explain fundamental chemical concepts including matter, atomic structure, molecules, bonding, physical and chemical properties, stoichiometry, acids and bases, chemical reactions and organic compounds
2. Interpret the periodic table of the elements
3. Apply chemical concepts and calculations of quantities to solve chemistry problems.

## Alignment of Learning Outcomes, Assessment and Graduate Attributes



### Alignment of Assessment Tasks to Learning Outcomes

Assessment Tasks	Learning Outcomes		
	1	2	3
1 - Online Quiz(zes) - 20%	•	•	
2 - Written Assessment - 30%	•	•	
3 - Written Assessment - 50%	•		•

### Alignment of Graduate Attributes to Learning Outcomes

Graduate Attributes	Learning Outcomes		
	1	2	3
1 - Self Management	—	—	—

Graduate Attributes	Learning Outcomes		
	1	2	3
2 - Communication	—		—
3 - Information Literacy	—	—	—
4 - Information Technology Competence	—		
5 - Problem Solving			—
6 - Critical Thinking			
7 - Cross-Cultural Competence			
8 - Ethical Practice			
9 - Aboriginal and Torres Strait Islander Cultures			

### Alignment of Assessment Tasks to Graduate Attributes

Assessment Tasks	Graduate Attributes								
	1	2	3	4	5	6	7	8	9
1 - Online Quiz(zes) - 20%	—			—					
2 - Written Assessment - 30%	—	—	—						
3 - Written Assessment - 50%	—	—	—		—				

## Textbooks and Resources

### Textbooks

CHEM40079

#### Prescribed

##### Introductory Chemistry

Edition: Third (2021)

Authors: School of Access Education

CQUniversity Publishing Unit

Rockhampton, Queensland, Australia

Binding: Spiral

#### Additional Textbook Information

A current printable version of the study guide can be found on the CHEM40079 Moodle site. The study guide is not available for purchase from the CQUniversity Bookshop, so it is strongly recommended that you print your own copy. Your Access Coordinator will provide advice on printing options.

Please note that the study guide also contains pictures and sections of text in colour which will not be shown in black and white prints. It is strongly advised that a digital version be saved to view these sections in colour.

### IT Resources

#### You will need access to the following IT resources:

- CQUniversity Student Email
- Internet
- Unit Website (Moodle)

## Referencing Style

All submissions for this unit must use the referencing style: [Harvard \(author-date\)](#)

For further information, see the Assessment Tasks.

## Teaching Contacts

**Flavia Santamaria** Unit Coordinator

[f.santamaria@cqu.edu.au](mailto:f.santamaria@cqu.edu.au)

## Schedule

### Week 1 - 11 Jul 2022

Module/Topic	Chapter	Events and Submissions/Topic
<b>Module 1:</b> Matter	<b>1</b>	Complete Week 1 practice quiz

### Week 2 - 18 Jul 2022

Module/Topic	Chapter	Events and Submissions/Topic
<b>Module 2:</b> Atoms and the Periodic Table	<b>2</b>	Complete week 2 practice quiz

### Week 3 - 25 Jul 2022

Module/Topic	Chapter	Events and Submissions/Topic
<b>Module 3:</b> Molecules and Ions	<b>3</b>	Assessment 1 (Online Quiz) available this week Complete week 3 practice quiz

### Week 4 - 01 Aug 2022

Module/Topic	Chapter	Events and Submissions/Topic
<b>Module 4:</b> Ionic and Covalent Compounds	<b>4</b>	Complete Week 4 practice quiz <b>Assessment 1 (Online Quiz)</b> Due: Week 4 Thursday (4 Aug 2022) 11:59 pm AEST
<b>Week 5 - 08 Aug 2022</b>		
<b>Module 5:</b> Chemical Reactions	<b>5</b>	Complete Week 5 practice quiz
<b>Vacation Week - 15 Aug 2022</b>		
<b>Study week</b>		Revision of topics
<b>Week 6 - 22 Aug 2022</b>		
<b>Module 6:</b> Redox Reactions	<b>6</b>	Complete Week 6 practice quiz
<b>Week 7 - 29 Aug 2022</b>		
<b>Module 7.</b> The Mole	<b>7</b>	Download Assessment 2 Complete Week 7 practice quiz
<b>Week 8 - 05 Sep 2022</b>		
<b>Module 8:</b> Concentration / Molarity	<b>8</b>	Complete Week 8 practice quiz <b>Assessment 2</b> Due: Week 8 Thursday (8 Sept 2022) 11:59 pm AEST
<b>Week 9 - 12 Sep 2022</b>		
<b>Module 9.</b> Reactions and calculations	<b>9</b>	Complete Week 9 practice quiz
<b>Week 10 - 19 Sep 2022</b>		
<b>Module 10.</b> Acids and bases	<b>10</b>	Complete Week 10 practice quiz
<b>Week 11 - 26 Sep 2022</b>		
<b>Module 11.</b> Organic chemistry	<b>11</b>	<b>Download Assessment 3</b> Complete Week 10 practice quiz
<b>Week 12 - 03 Oct 2022</b>		
<b>Module 12.</b> Chemical equilibrium	<b>12</b>	<b>Assessment 3</b> Due: Week 12 Thursday (6 Oct 2022) 11:59 pm AEST
<b>Review/Exam Week - 10 Oct 2022</b>		
<b>Module/Topic</b>	<b>Chapter</b>	<b>Events and Submissions/Topic</b>
<b>Exam Week - 17 Oct 2022</b>		
<b>Module/Topic</b>	<b>Chapter</b>	<b>Events and Submissions/Topic</b>

## Term Specific Information

For any assistance and to organise one-on-one or group Zoom sessions please contact:  
Unit Coordinator and Lecturer Dr Flavia Santamaria  
Email: f.santamaria@cqu.edu.au  
Ph: (07) 4930 6978.

## Assessment Tasks

### 1 Assessment 1 (Online Quiz)

**Assessment Type**

Online Quiz(zes)

**Task Description**

Assessment 1 is available on the CHEM40079 Moodle site. It is a compulsory online Quiz that covers material from Modules 1, 2 and 3. Marks will be awarded for the correct response to multiple choice and short answer questions. Once you start the Quiz, you have 3.5 hours to complete the 30 questions.

Please note the following requirements and advice:

- Follow the instructions provided throughout the quiz.
- Do not use the internet in seeking answers; this is not a research task and internet resources may be incorrect.
- Paraphrase (use your own words) the information found in the Study Guide to demonstrate your understanding of topics.
- Use logical explanations and problem-solving.
- Use correct chemical formulas including appropriate subscripts and superscripts.

The resources provided on the CHEM40079 Moodle site contain all the relevant material (content, examples and practice quizzes) required to complete the assessment.

**Number of Quizzes**

1

**Frequency of Quizzes**

Other

**Assessment Due Date**

Week 4 Thursday (4 Aug 2022) 11:59 pm AEST

This quiz closes at 11:59 pm. Ensure to complete and submit the quiz by this time. After this time, the quiz will only be available to students with an extension.

**Return Date to Students**

Week 6 Thursday (25 Aug 2022)

This assessment will be returned within two weeks of the assessment due date or received date, whichever is later.

**Weighting**

20%

**Assessment Criteria**

Marks will be awarded for the correct response to multiple choice and short answer questions. Responses are scored based on:

- correct use of terminology
- factual correctness of presented material
- correct use of science conventions
- calculations which should include units where relevant
- relevance of stated content to the question asked
- application of foundation concepts to the question asked
- clarity, thoroughness and completeness of explanations
- logic of explanations and problem-solving
- correct use of chemical formulas including appropriate subscripts and superscripts.

## Referencing Style

- [Harvard \(author-date\)](#)

## Submission

Online

## Learning Outcomes Assessed

- Explain fundamental chemical concepts including matter, atomic structure, molecules, bonding, physical and chemical properties, stoichiometry, acids and bases, chemical reactions and organic compounds
- Interpret the periodic table of the elements

## Graduate Attributes

- Self Management
- Information Technology Competence

## 2 Assessment 2

### Assessment Type

Written Assessment

### Task Description

Assessment 2 is available on the CHEM40079 Moodle site. It is a compulsory written assessment that covers material from Modules 4, 5 6 and 7, in addition to fundamental concepts covered in earlier modules.. The assessment comprises various question types that require handwritten responses.

Depending on the question, you may be required to do one or more of the following:

- Perform calculations, providing units where relevant.
- Balance equations including states of matter.
- Present information graphically.
- Use of examples and diagrams to support your answer.
- Write chemical formulas using appropriate subscripts and superscripts.

### Assessment Due Date

Week 8 Thursday (8 Sept 2022) 11:59 pm AEST

### Return Date to Students

Week 10 Thursday (22 Sept 2022)

This assessment will be returned within two weeks of the assessment due date or received date, whichever is later.

### Weighting

30%

### Assessment Criteria

Marks are allocated for the following:

- appropriate setting of the solution
- correct sequence of steps
- relevant use of examples and diagrams
- accurate use of units, equations and formulas

The number of marks allocated to each question is detailed in the assessment task on Moodle. Answers to all questions must be handwritten on the assessment task sheets, using additional paper if extra space is required, and clearly presented with full working provided in order to obtain the maximum allocation of marks.

## Referencing Style

- [Harvard \(author-date\)](#)

## Submission

Online

## Submission Instructions

Assessment 2 is to be uploaded as a single pdf document through the CHEM40079 Moodle site.

## Learning Outcomes Assessed

- Explain fundamental chemical concepts including matter, atomic structure, molecules, bonding, physical and chemical properties, stoichiometry, acids and bases, chemical reactions and organic compounds
- Interpret the periodic table of the elements



## Graduate Attributes

- Self Management
- Communication
- Information Literacy

## 3 Assessment 3

### Assessment Type

Written Assessment

### Task Description

Assessment 3 is available via the CHEM40079 Moodle site. It is a compulsory written assessment that covers material from Modules 8, 9, 10 and 11, in addition to fundamental concepts covered in earlier modules. The assessment comprises various question types that require handwritten responses. Depending on the question, you may be required to do one or more of the following:

- Choose the correct response from multiple-choice questions.
- Perform simple or multi-step calculations, providing and using appropriate formulas and units where relevant.
- Analyse primary and secondary data to determine solutions.
- Provide short or extended answer responses using examples and diagrams to support your answer.
- Present information in the correct chemical format (i.e. use subscripts and superscripts, and balance equations, including states of matter).

### Please note the following requirements and advice:

- The resources provided on the CHEM40079 Moodle site contain all the relevant material (content, examples and practice quizzes) required to complete the assessment. All working, including the development of answers, must be shown.
- Ensure that you do not use the internet in seeking answers, as this assessment is not a research task, but is assessing your knowledge and understanding of topics covered in the Unit. Moreover, the information from some sites is unreliable, or not as specific as is required for this assessment.
- Use your own words (paraphrase) the information found in the Study Guide rather than copying it word for word. Conveying the information in your own words will demonstrate your understanding of topics

### Assessment Due Date

Week 12 Thursday (6 Oct 2022) 11:59 pm AEST

### Return Date to Students

This assessment will be returned within two weeks of the assessment due date or received date, whichever is later.

### Weighting

50%

### Minimum mark or grade

To pass this Unit, you must obtain at least 50% of the marks available for this assessment.

### Assessment Criteria

Marks are allocated for the following:

- appropriate setting of the solution
- correct sequence of steps
- relevant use of examples and diagrams
- accurate use of units, equations and formulas

The number of marks allocated to each question is detailed in the assessment task on Moodle. Answers to all questions must be handwritten on the assessment task sheets, using additional paper if extra space is required, and clearly presented with full working provided in order to obtain the maximum allocation of marks.

### Referencing Style

- [Harvard \(author-date\)](#)

### Submission

Online

### Submission Instructions

Assessment 3 is to be uploaded as a single pdf document through the CHEM40079 Moodle site.

## Learning Outcomes Assessed

- Explain fundamental chemical concepts including matter, atomic structure, molecules, bonding, physical and chemical properties, stoichiometry, acids and bases, chemical reactions and organic compounds
- Apply chemical concepts and calculations of quantities to solve chemistry problems.

## Graduate Attributes

- Self Management
- Communication
- Information Literacy
- Problem Solving

## Academic Integrity Statement

As a CQUniversity student you are expected to act honestly in all aspects of your academic work.

Any assessable work undertaken or submitted for review or assessment must be your own work. Assessable work is any type of work you do to meet the assessment requirements in the unit, including draft work submitted for review and feedback and final work to be assessed.

When you use the ideas, words or data of others in your assessment, you must thoroughly and clearly acknowledge the source of this information by using the correct referencing style for your unit. Using others' work without proper acknowledgement may be considered a form of intellectual dishonesty.

Participating honestly, respectfully, responsibly, and fairly in your university study ensures the CQUniversity qualification you earn will be valued as a true indication of your individual academic achievement and will continue to receive the respect and recognition it deserves.

As a student, you are responsible for reading and following CQUniversity's policies, including the [Student Academic Integrity Policy and Procedure](#). This policy sets out CQUniversity's expectations of you to act with integrity, examples of academic integrity breaches to avoid, the processes used to address alleged breaches of academic integrity, and potential penalties.

### What is a breach of academic integrity?

A breach of academic integrity includes but is not limited to plagiarism, self-plagiarism, collusion, cheating, contract cheating, and academic misconduct. The Student Academic Integrity Policy and Procedure defines what these terms mean and gives examples.

### Why is academic integrity important?

A breach of academic integrity may result in one or more penalties, including suspension or even expulsion from the University. It can also have negative implications for student visas and future enrolment at CQUniversity or elsewhere. Students who engage in contract cheating also risk being blackmailed by contract cheating services.

### Where can I get assistance?

For academic advice and guidance, the [Academic Learning Centre \(ALC\)](#) can support you in becoming confident in completing assessments with integrity and of high standard.

### What can you do to act with integrity?



### **Be Honest**

If your assessment task is done by someone else, it would be dishonest of you to claim it as your own



### **Seek Help**

If you are not sure about how to cite or reference in essays, reports etc, then seek help from your lecturer, the library or the Academic Learning Centre (ALC)



### **Produce Original Work**

Originality comes from your ability to read widely, think critically, and apply your gained knowledge to address a question or problem